

In Situ Functionalized Fluorescent WS₂-QDs as Sensitive and Selective Probe for Fe³⁺ and a Detailed Study of Its Fluorescence Quenching

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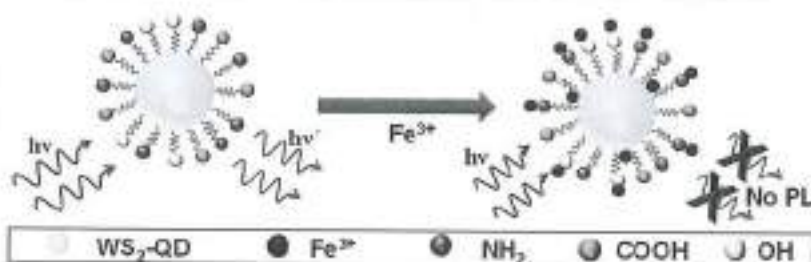
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Supporting Information



ABSTRACT: Most of the reports suggest that liquid exfoliated WS₂-QDs are unstable; therefore the need of present day is to develop a novel synthesis route for producing long-term stable WS₂-QDs. Herein, we report a bottom-up single-step hydrothermal growth of in situ functionalized blue fluorescent WS₂-QDs with stable fluorescence in aqueous media without subsequent treatments. Presence of various functional groups over the surface of f-WS₂-QDs provides high solubility and stability to f-WS₂-QDs in aqueous media preserving its fluorescence. Further, photoluminescence property of f-WS₂-QDs has been employed to devise an optical sensor with a high sensitivity ($K_D \sim 10^4 \text{ M}^{-1}$) and selectivity for ferric (Fe³⁺) ions. Under the optimal condition, response of the sensor is found to be linear in the range of 0–55 μM with a limit of detection (LOD) of 1.32 μM , which is within the maximum permissible level of Fe³⁺ ($\sim 5.4 \mu\text{M}$) in human drinking water by the USEPA. Further, we have also carried out a detailed evaluation on fluorescence quenching kinetics of f-WS₂-QDs. Nonlinear behavior of S–V plot and TRPL measurements suggest that quenching is a mixed phenomenon of dynamic as well as static processes. Finally we have proposed a mechanism for fluorescence quenching of f-WS₂-QDs in the presence of Fe³⁺.

KEYWORDS: WS₂, quantum dots, hydrothermal synthesis, quantum yield, TRPL, fluorescence quenching

INTRODUCTION

In the post-graphene era, a great sensation has been observed in the current research on layered transition metal dichalcogenides (LTMDs) of few-to-monolayer nanocrystals (NCs) or quantum dots (QDs) of MoS₂, WS₂, MoSe₂, etc., which have been conceived as best substitutes of graphene and carbon QDs.^{1–4} In this LTMDs series, 2D WS₂ has aroused even greater attention owing to its controllable band gaps as a function of layer numbers with remarkable optical and electronic properties.^{5–7} WS₂ is a van der Waals layered material with hexagonal lattice structure arranged in triple layers (S–W–S), each layer has a thickness of 6 Å with strong

in-plane covalent bonding and weak out-of-plane van der Waals interactions.⁸ Further, by reducing the lateral size of 2D WS₂ sheet in 0D WS₂-QDs, one can acquire excellent electrical/optical behaviors from these NCs due to their strong quantum confinement and edge effects.⁹ In addition to this, WS₂-QDs possess larger surface to volume ratio and much more active sites beyond monolayer 2D WS₂, which may be beneficial for sensing, catalysis, bioimaging, etc. To date,

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