

Highly efficient infrared to visible up-conversion emission tuning from red to white in Eu/Yb co-doped NaYF₄ phosphorTrilok K. Pathak^{a,b,*}, Ashwini Kumar^{a,†}, L.J.B. Erasmus^a, Anurag Pandey^a, E. Coetsee^a, H.C. Swart^a, R.E. Kroon^{a,††}^a Department of Physics, University of the Free State, Bloemfontein, South Africa
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ABSTRACT

Eu³⁺/Yb³⁺ co-doped NaYF₄ phosphors have been synthesized by the combustion method. The Eu³⁺ doping level and the effect of Yb³⁺ doping concentration on the structural, morphological and luminescence properties have been investigated. X-ray diffraction analysis revealed that the phosphor consisted of mixed α- and β-phases, but the β-phase was dominant. All elements of the host and dopants, as well as adsorption sites, were detected using X-ray photoelectron spectroscopy. The surface morphology showed a nanorod-like structure with sharp hexagonal edges. Energy dispersive X-ray spectroscopy spectra proved the formation of the desired material. The photoluminescence spectra illustrated the up-conversion spectra of Eu³⁺ in the red region when excited at 980 nm, while, under the same excitation, Yb³⁺ fluorescence emission at 980 nm. The up-conversion (UC) emission of Eu³⁺/Yb³⁺ co-doped NaYF₄ produced a white color at the higher concentrations of Yb³⁺ excited by a 980 nm laser, which was made possible by more emission from Yb³⁺ ions and less emission of Eu³⁺ ions. The lifetime of the Eu³⁺ UC luminescence at 615 nm was also affected by the Yb³⁺ doping concentration. The temperature sensitivity associated with the Eu³⁺ peaks at 520 and 542 nm was analyzed as a function of temperature and the maximum of 0.0463 K⁻¹ was observed at 400 K.

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1. Introduction

The study of phosphors which emit visible light upon sequential absorption of two or more low energy (near infrared, NIR) photons led the foundation for the new field of up-conversion (UC) nanocrystals [1,2]. This phenomenon, termed UC luminescence, has many potential applications such as solar cells, pH sensors and medical imaging [3–6]. There is a constant demand on developing a white UC phosphor with high efficiency. This work was done to synthesize nanocrystalline Eu³⁺/Yb³⁺ doped NaYF₄ using the combustion method and to investigate the effect of Yb³⁺ concentration on its structure, morphology and luminescence properties (including color). Fluoride materials normally possess a low photon energy and high refractive index. Furthermore, they exhibit adequate thermal and environmental stability. Therefore, fluoride materials are regarded as excellent host lattices for down-conversion and UC luminescence of lanthanide ions [7–11]. Among the investigated fluorides, NaYF₄ has attracted increasing attention in the field of

materials science over the past two decades [12–14]. It has three possible modifications: (i) the low-temperature cubic (α) form, (ii) the hexagonal (β) form and (iii) the high-temperature cubic form, depending on the synthesis conditions and methods [15,16]. Zhao et al. [17] observed highest efficiency for UC luminescence from hexagonal NaYF₄/Yb³⁺ material, mainly due to the low photon energy of the NaYF₄ lattice. The rare earth combination Yb³⁺/Eu³⁺ has mostly been used to show UC phenomena in different host lattices, whereas few studies have been reported with the Yb³⁺/Eu³⁺ doping combination. However, Eu³⁺ ions have excellent red emission properties and are famous structural probes [18–24]. Wang et al. [25] observed not only the white UC luminescence, but also the ultra-broadband red UC fluorescence of Eu³⁺ in YF₃ nanocrystals by using Yb³⁺ and Th⁴⁺ as sensitizers and a 980 nm diode laser as a pump source. In the present study, Eu³⁺/Yb³⁺ doped NaYF₄ was synthesized and its UC luminescence and lifetime properties at excited by a 980 nm diode laser were investigated as a function of Yb³⁺ concentration, showing that it can act as a white UC phosphor. We have also studied its temperature sensitivity as a function of temperature.

2. Experimental details

Analytic grade yttrium nitrate hexahydrate Y(NO₃)₃·6H₂O (99.99%), sodium nitrate NaNO₃ (99.99%), ammonium fluoride NH₄F (99.99%),

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